First Enantiocontrolled Formal Synthesis of (+)-Neovibsanin B, A Neurotrophic Diterpenoid

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ABSTRACT



An enantiocontrolled formal synthesis of (+)-neovibsanin B has been achieved by a sequence that applies an asymmetric 1,4-addition of $(H_2C=CH)_2Cu(CN)Li_2$ to trisubstituted $\alpha_{,\beta}$ -carboxylic acid derivative 1 to induce the chirality at the C-11 all-carbon quaternary center. Together with a modified Negishi cyclic carbopalladation-carbonylative esterification tandem reaction for constructing the A-ring, the synthesis was completed.

Neovibsanins A and B, vibsane-type diterpenoids, which were isolated from the leaves of *Viburnum awabuki* by Fukuyama et al.,¹ have attracted considerable synthetic attention because of their challenging structures combined with interesting neurotrophic activity. They have been found to significantly promote the neurite outgrowth of NGF-mediated PC12 cells,² and thus have shown potential as drugs for the treatment of neurodegenerative diseases such as Alzheimer's disease.³ Recently, Nishizawa et al.⁴ achieved the total synthesis of (\pm) -neovibsanin B by utilizing an intramolecular Diels–Alder reaction to construct the A-ring system and a chelation-controlled diastereoselective alkylation at C-4 as two key steps, and the resultant synthetic (\pm) -neovibsanin B was shown to exhibit neurite outgrowth-promoting activity in NGF-mediated PC12 cells comparable

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to that of natural (+)-neovibsanin B. On the other hand, Williams et al.⁵ reported a synthesis of (\pm)-4,5-bis-*epi*-neovibsanins A and B, which were also shown to accelerate neurite outgrowth in NGF-mediated PC12 cells (Figure 1).



Figure 1. Structures of neovibsanins A and B.

However, no enantioselective synthetic study on neovibsanins has been published. Herein, we wish to demonstrate the first enantiocontrolled formal synthesis of (+)-neovibsanin B

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Scheme 1. Synthetic Plan of (+)-Neovibsanin B



based on an asymmetric 1,4-addition, which was previously developed by us,⁶ and a modified Negishi cyclic carbopalladation-carbonylative esterification tandem reaction.⁷

Our synthetic plan of (+)-neovibsanin B is outlined in Scheme 1. To achieve its synthesis in an enantiocontrolled fashion, the enantioselective construction of the chiral all carbon quaternary center at C-11 was a requirement. We decided to employ the asymmetric 1,4-addition reaction of $(H_2C=CH)_2Cu(CN)Li_2$ to the trisubstituted α,β -carboxylic acid derivative 1 of (R)-4-phenyl-2-oxazolidinone⁶ to prepare (11S)-2. According to our previous report,⁸ removal of the chiral auxiliary of (11S)-2 followed by the alkynylation and selective iodination would lead to iodoalkene, (11S)-3. Previously we reported that the modified Negishi palladium(0)-catalyzed carbonylative cyclization⁷ of (\pm) -3 smoothly proceeded to give rise to the cyclohexen-1-one derivative, (\pm) -4.⁸ Therefore, (11S)-4 would be prepared from (11S)-3 by employing the same reaction (Scheme 2). After C1 extension from the exo-methylene group by a 1,4addition reaction, simple functional group manipulation would lead to Nishizawa's intermediate, (11S)-5 in optical active form.

Negishi's iodomethylation⁹ of 4-pentyn-1-ol gave the alkenyl iodide 6, which was followed by the silylation of

the hydroxy group with TBDPSCl. Subsequent palladium(0)catalyzed carbonylation-amidation of 6 with (R)-4-phenyl-2-oxazolidinone afforded the trisubstituted α,β -carboxylic acid derivative 1 in 68% yield. The asymmetric 1,4-addition reaction of $(H_2C=CH)_2Cu(CN)Li_2$ to 1 gave rise to 2 as a diastereomeric mixture of 95 (11S):5 (11R) in good yield, and each diastereomer was readily separated by column chromatography over silica gel. The optically pure (11S)isomer of 2 was used for the subsequent reaction. Removal of the oxazolidinone in 2 with 30% aqueous $H_2O_2/LiOH^{10}$ followed by esterification with EtOH yielded the ester, which was then reduced with LiAlH₄ to the alcohol 7 in 87% yield over three steps. PCC oxidation of 7, followed by Corey-Fuchs dibromoolefination,¹¹ provided 1,1-dibromoolefin 8 in 89% yield. The stereoselective conversion of 8 into the cyclization precursor (2Z)-3 was achieved by the following sequence. Treatment of 8 with *n*-BuLi generated the alkynyl anion, which was trapped in situ with formaldehyde, giving rise to the alkynyl alcohol. Regio- and stereoselective hydrostannylation¹² of its alkyne moiety with tributyltin hydride in the presence of AIBN exclusively afforded (2Z,11S)-3 after protecting the hydroxyl group as a TBS group in 83% yield over four steps. Then, carbonylative cyclization of (2Z)-3 in the presence of 10 mol % of



Table 1. Negishi's Pd(0)-Catalyzed Cyclic Carbopalladation-Carbonylative Tandem Reaction of (2Z,11S)-3

	OTBDPS (2Z,115)-3	MeO nol % PdCl ₂ (PPh ₃) ₂ 36 hrs	O O Me OTBS + (10 <i>R</i> ,11 <i>S</i>)-9	0 	DTBS + MeO BDPS 10	O OTBS +		[∼] OTBS)TBDPS	
entry	base (1.5 equiv)	solvent	MeOH (equiv)	CO (MPa)	temp. (°C)	$9 (10R:10S)^a$	10	4	3 (%)
1	$\mathrm{Et}_3\mathrm{N}$	MeCN/PhH (1:1)	4	4	100	11 (1.1:1)	0	6	52
2	$\mathrm{Et}_{3}\mathrm{N}$	MeCN/PhH (1:1)	48	4	100	54 (2.4:1)	10	2	0
3	$\mathrm{Et}_{3}\mathrm{N}$	MeCN/PhH (1:1)	48	8	100	41 (2.3:1)	9	12	0
4	$\mathrm{Et}_{3}\mathrm{N}$	MeCN/PhH (1:1)	48	4	60	49 (2.7:1)	13	0	21
5	$\mathrm{Et}_{3}\mathrm{N}$	MeCN/PhH (1:1)	24	4	60	69 (2.6:1)	14	0	0
6	$\mathrm{Et}_{3}\mathrm{N}$	1,4-dioxane	4	4	100	0	0	0	90
7	$\mathrm{Et}_{3}\mathrm{N}$	MeOH	_	4	100	6 (1.4:1)	13	4	52
8	K_3PO_4	MeCN/PhH (1:1)	4	4	100	16 (1.4:1)	18	0	3
9	i - Pr_2NEt	MeCN/PhH (1:1)	4	4	100	0	0	0	85
10	DABCO	MeCN/PhH (1:1)	4	4	100	24(1.6:1)	0	9	0
^a Ratio was determined by ¹ H NMR spectroscopy in CDCl ₃ (300 MHz).									

PdCl₂(PPh₃)₂ under a carbon monoxide atmosphere (0.4 MPa) smoothly proceeded to give rise to the desired cyclohexenone derivative (11*S*)-4 in 56% yield. In order to produce (10*R*,11*S*)-5 from (11*S*)-4, we tried the C1-extension from the *exo*-methylene moiety of (11*S*)-4 by using 1,4-addition reaction of a CN⁻ species. However, all these attempts were fruitless due to the poor electrophilicity of the conjugated *exo*-methylene moiety, which was diminished by the additional conjugated endocyclic olefin. This result prompted us to explore alternative routes to the key intermediate **5** with an extended C1 unit.

In principle, compound **3** could be obtained by Negishi palladium(0)-catalyzed cyclic carbopalladation-carbonylative tandem reaction in the presence of MeOH.^{7b} The desired product **9**, which has a methyl ester group, would be directly formed instead of the *exo*-methylene type compound **4**. Thus, we examined this reaction. First, the reaction was performed using 5 mol % PdCl₂(PPh₃)₂ and Et₃N (1.5 equiv) in MeCN/PhH (1:1) containing 4 equiv of MeOH at 100 °C in an autoclave, which gave rise to the desired diastereomeric mixture of **9** in 11% yield along with ca. 50% of the starting material containing a small amount of **4** (6%) (Table 1, entry

1). On the other hand, the addition of an excess amount (48 equiv) of MeOH to this reaction system dramatically increased the yield of 9 to 54%, contaminated with 10% of the noncyclic ester 10 (entry 2). The use of high pressure (8 MPa) was found to be ineffective at suppressing the generation of 10 (entry 3), but low temperature (60 $^{\circ}$ C) was able to decrease the generation of 4 (entry 4). After several trials, we were pleased to find that the following reaction conditions of 24 equiv of MeOH, 4 MPa CO and a temperature of 60 °C, led to the formation of 9 alone in ca. 70% yield as a diastereometric mixture (10R:10S = 2.6:1)(entry 5). Each diastereomer of 9 was readily separated by silica gel column chromatography to give (10R,11S)-9 and (10S,11S)-9. Furthermore, treatment of the undesired isomer, (10S,11S)-9 with MeOLi in MeOH gave the equilibrating diastereomers (10R,11S)-9 and (10S,11S)-9 in a ratio of 1.5: 1. Thus, repeating this operation enabled the conversion of the undesired isomer (10S, 11S)-9 to the desired isomer (10R,11S)-9. This was due to the 1,3-diaxial interaction between the side chain at the C10 position and the proton at the C1 position as depicted in Scheme 3.





With (10R,11S)-9 in hand, we focused on the last few steps for the synthesis of Nishizawa's intermediate 5 (Scheme 4). Treatment of (10R,11S)-9 with *n*-Bu₄NF containing acetic acid gave 11, and the resultant hydroxy group was oxidized by Swern oxidation to its aldehyde, which was subjected to

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Scheme 4. Completion of Enantiocontrolled Formal Synthesis of (+)-Neovibsanin B



Wittig olefination to give the dimethyl olefin **12** in 69% yield over two steps. Reduction of **12** with DIBAL-H provided the cyclic hemiacetal **13** and diol **14** in 29 and 26% yields, respectively. The cyclic hemiacetal **13** was transformed to **14** by reduction with NaBH₄. Unfortunately, the selective protection of the primary alcohol in **14** using the reaction conditions, 2,4- DMPMCl and Bu₂SnO, utilized by Nishizawa et al.^{4,13} did not succeed in our hands. However, when **14** was reacted with freshly prepared 2,4-DMPM-trichloroacetoimidate¹⁴ in the presence of 10 mol % CSA, the desired 2,4-DMPM-ether **15** was obtained in 45% yield. Finally Dess-Martin oxidation¹⁵ of **15** afforded the Nishizawa's

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intermediate **5** as an optically active form ($[\alpha]_D^{22} + 20.1$ (*c* 1.05, MeOH)) in 95% yield. The spectroscopic data (MS, IR, ¹H NMR, ¹³C NMR) of our synthetic **5** were identical to those of Nishizawa's intermediate.⁴

In conclusion, the first enantiocontrolled formal synthesis of (+)-neovibsanin B was accomplished in 1.1% overall yield over 20 steps. Our formal synthesis of (+)neovibsanin B not only demonstrates a useful application of the asymmetric 1,4-addition of the trisubstituted α , β carboxylic acid derivative **1**, but also emphasizes the potential use of the Negishi's cyclic carbopalladationcarbonylative esterification tandem reaction for constructing the neovibsanin skeleton. Work is now in progress to achieve total synthesis of (+)-neovibsanin B.

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Supporting Information Available: Experimental procedures and characterization of products. This material is available free of charge via the Internet at http://pubs.acs.org.

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